

Jordan University of Science and Technology Faculty of Institute Of Nanotechnology Nanotechnology And Engineering Department

NANO708 Chemistry Of Nanomaterials - JNQF Level: 9

Second Semester 2023-2024

Course Catalog

3 Credit Hours. 3 Credit hours (3 h lectures). This course aims to provide a comprehensive overview of the chemical and physical properties and reactions of Nanomaterials and how they are different from bulk materials. Topics covered include the chemical composition of Nanomaterials, Nanocrystal, Clusters, Dendrimers, Nanodispersion, Nanoporous materials, Composites and Nanocomposites, types, properties, reactivity, preparation, and characterization. Emphasis is placed on understanding the design and properties of nanomaterials that are utilized in advanced applications, including catalysts (nanoparticles, Nanodispersion), carriers of drug delivery (Nanoparticles, Dendrimers), modified membranes (clay/polymer), lightweight high-strength materials (CNT/Polymer nanocomposite), biosensors (Dendrimers).

Teaching Method: On Campus

	Text Book			
Title	Nanomaterials and Nanochemistry			
Author(s)	C. Br?echignac P. Houdy M. Lahmani (Eds.)			
Edition	1st Edition			
Short Name	1			
Other Information	Springer/ 2007			

Course References

Short name	Book name	Author(s)	Edition	Other Information
2	Nanoclusters, Volume 1 A Bridge across Disciplines	Purusottam Jena, A. Welford Castleman Jr. Jr. (Eds.)	1st Edition	Elsevier/ 2010/Volume 1A
3	The Delivery of Nanoparticles	Edited by Dr. Abbass A. Hashim	1st Edition	2012
4	Nanomaterials Chemistry Recent Development and New Direction	Edited by C. N. R. Rao, A. M?ller, A. K. Cheetham	1st Edition	2007

5	The Chemistry of Nanomaterials, Synthesis, Properties and Application	C. N. R. Rao, A. Muller, A. K. Cheetham	1st Edition	Two volumes
6	Essentials in Nanoscience and nanotechnology	Kumar, Narendra & Kumbhat, Sunita	1st Edition	2016/ John Wiley & Sons
7	Handouts	Dr. Ayat Bozeya	1st Edition	

Instructor			
Name	Dr. Ayat Bouzieh		
Office Location	Institute of Nanotechnology		
Office Hours	Sun : 10:30 - 11:15 Mon : 12:30 - 14:00 Tue : 10:45 - 11:45 Thu : 10:30 - 13:30		
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Class Schedule & Room

Section 1: Lecture Time: Tue : 13:30 - 16:30 Room: NANO 1

Tentative List of Topics Covered				
Weeks	Торіс	References		
Week 1	Introduction to nanomaterials in Chemistry	Ch. # 1 From 1, Ch. # 1 From 5, L1 From 7		
Week 2	Nanocrystal/ Reactivity of Colloidal Inorganic Nanoparticles	Ch. # 18 From 1, Ch. # 18 From 3, Ch. # 1 From 4, L2 From 7		
Week 3	Structure and Phase Transitions in Nanocrystals	Ch. # 2 From 1 , L3 From 7		
Week 4	Clusters	Ch. # 6 From 1 , Ch. # 1 From 2 , L4 From 7		
Week 5	Nanodispersion	Ch # 11. & Ch. # 12 From 1, L5 From 7		
Week 6	Midterm Exam	L1, L2, L3, L4, L5 From 7		
Weeks 7, 8	Synthetic Methods for nanomaterials / Chemistry aspects	Ch. # 15, Ch. # 16, Ch. # 17 From 1, L6, L7 From 7		

Weeks 9, 10	Dendrimers	Ch. #18 From 3 , Ch. # 8 From 4 , L8, L9 From 7
Weeks 11, 12	Nano porous Solid Materials	Ch. # 26 From 1, Ch. # 18 From 5, L10, L11 From 7
Weeks 13, 14	Composites & Nanocomposites	Ch. # 14 From 1 , Ch. # 7 From 6 , L12, L13 From 7
Week 15	Selected Application of Nanomaterials	L14 From 7
Week 16	Final Exam	L6 - L13 From 7

Mapping of Course Outcomes to Program Outcomes and NQF Outcomes	Course Outcome Weight (Out of 100%)	Assessment method
By the end of the course, the students are expected to Explain the differences between bulk and nanomaterial properties [5PLO1] [5L9K1]	5%	
By the end of the course, the students are expected to explain the effects of quantum confinement on the electronic structure and corresponding physical and chemical properties of materials at the nanoscale. [7PLO1] [7L9K1]	7%	
By the end of the course, the students are expected to achieve the ability to choose appropriate synthesis techniques to synthesize different nanomaterials of desired properties: size, shape, and surface properties. [15PLO3] [15L9K3]	15%	
By the end of the course, the students are expected to correlate properties of nanomaterials with their size, shape, and surface characteristics. [8PLO3] [8L9K3]	8%	
By the end of the course, the students are expected to appreciate the enhanced sensitivity of nanomaterial-based sensors and their novel applications. [11PLO4] [6L9S1, 5L9C2]	11%	
By the end of the course, the students are expected to explain the synthesis of such nanomaterials from chemistry aspects. [12PLO4] [6L9S1, 6L9C2]	12%	
By the end of the course, the students are expected to be able to discuss the benefits and use of dendrimers [14PLO5] [10L9S2, 4L9C1]	14%	
By the end of the course, the students are expected to be able to discuss the benefit and use of Nanoporous materials [14PLO5] [8L9S2, 6L9C1]	14%	
By the end of the course, the students are expected to be able to discuss the benefit and use of Nanocomposite [1PLO5] [7L9S2, 7L9C1]	14%	

		Relations	nip to Progra	m Student O	utcomes (Ou	ıt of 100%)		
PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9
12		23	23	42				

Relationship to NQF Outcomes (Out of 100%)					
L9K1	L9K3	L9S1	L9S2	L9C1	L9C2
12	23	12	25	17	11

Evaluation			
Assessment Tool	Weight		
Midterm exam	35%		
Final Exam	50%		
Homework & Quizzes	5%		
Term Paper	10%		

	Policy
Exams	 Applied project and term papers are required for graduate students. Plagiarism laws will be enforced on all class assignments, papers, projects, and tasks. Exams Could be a mix of (Short essays, short answers, multiple choice, fill-in-the-blank, matching, etc.). They cover all lectures including all readings, lectures, handout material, and postings on the E-Learning module. Exam makeup is accepted only for emergencies. Students absent from an examination must report immediately to the graduate registrar?s office following JUST regulation. A student who is absent on account of illness must submit a medical certificate evidence. No make-up for quizzes During exams You are not allowed to access or consult any materials, resources, websites, or applications (both hardcopy and digital) that your examiner has not provided as part of the exam questions and any accompanying appendices (where applicable). You are not allowed to communicate with another student during the exam, nor look at another?s answers. You may not leave the exam until the midway point of the examination time. You must ONLY use blue or black pens (no pencils). You must provide your materials (pens, calculators, etc.). No spare materials will be provided, and no material in an exam may be shared. Late students are allowed to enter the exam room during the first half midway point of the examination time, but they will not be given any extra time.
Cheating	Prohibited; The commitment of the acts of cheating and deceit such as copying during examinations, altering examinations for re-grade, plagiarism of homework assignments, and in any way representing the work of others as your own is dishonest and will not be tolerated. Standard JUST policy will be applied. Induces 7: إذا ضُبط الطالب أثناء الامتحان أو الاختبار متلبساً بالغش فتوقع عليه العقوبات التالية مجتمعة . أ- اعتبار ه راسباً في ذلك الامتحان أو الاختبار متلبساً بالغش فتوقع عليه العقوبات التالية مجتمعة . ب- الغاء تسجيله في بقية المساقات المسجل لها في ذلك الفصل . ج- فصله من الجامعة لمدة فصل در اسي واحد، و هو الفصل التالي للفصل الذي ضبط فيه
Attendance	 Attendance is mandatory and will be recorded regularly. Excellent attendance is expected. Students who miss more than 20% of the classes will be dropped from the course as per JUST policy. If you miss class, it is your responsibility to find out about any announcements or assignments you may have missed.

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